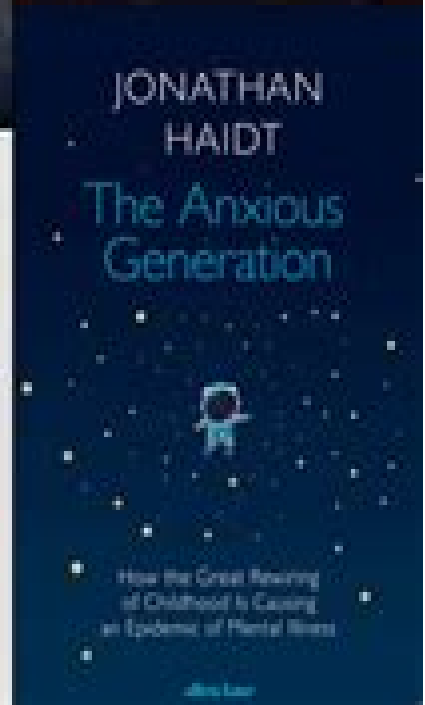
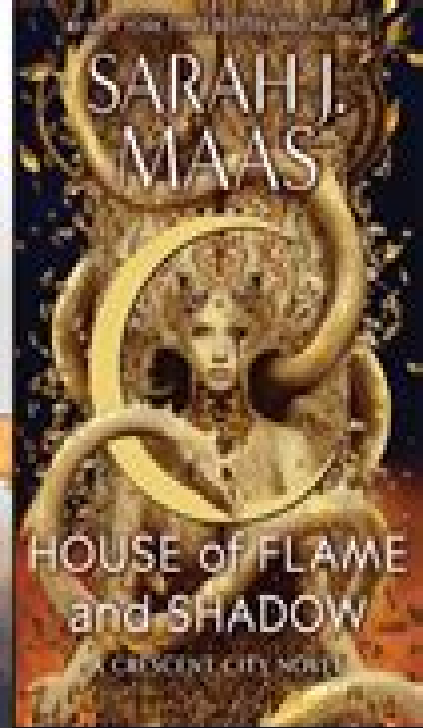
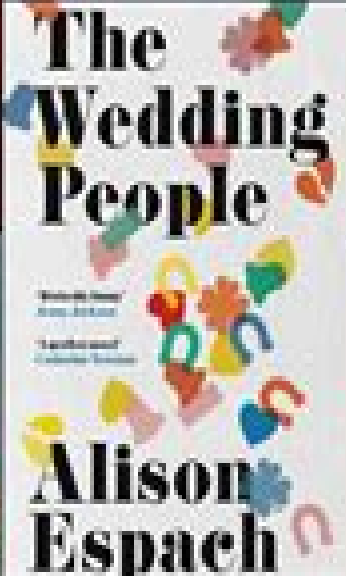
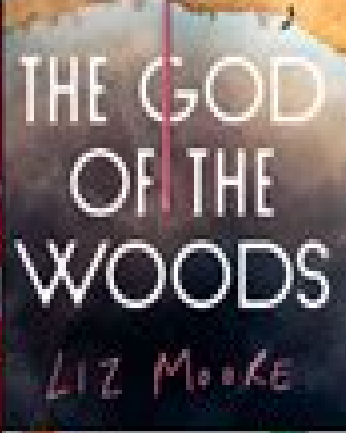
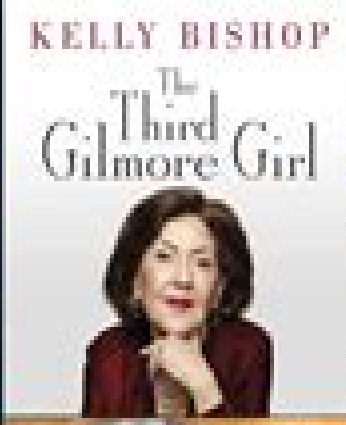




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$$\text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$$
construct a table showing how to interpret the information in this equation in terms of a single molecule of glucose moles of reactants and products

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web there's 2 on the left and 1 on the right so we need to change the coefficient of NH_3 to 2 now we have $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ total the atoms up again on the left there is 2 N and 6 H still on the right there is 2 N and 6 H now so now all we need to do is make the left side have 6 H in total

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