

STUDY GUIDE 11: Oxidation and Reduction (Chapter 12)

Remember: Oxidation Cannot Occur Alone!

Oxidation: loss of electrons, decrease in O.N. number (including 0,0 bonds)

Reduction: gain of electrons, increase in O.N. number, (decreasing 0,0 bonds)



Be careful: some compounds can contain both oxidized and reduced carbons, but the overall reaction is still an oxidation or reduction.



PROBLEM 12.14: ALKYLATION Reduction with LiAlH_4

Aluminum

Aluminum is a very electropositive metal (usually Al^{+3} , Al^{+2} , or Al^{+1}) which makes it a good catalyst.



Procedure: Perfect alcohols (isopropanol) is produced by using these compounds which apply the catalyst catalyst



Oxidation And Reduction Study Guide

Albert A Gayle



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