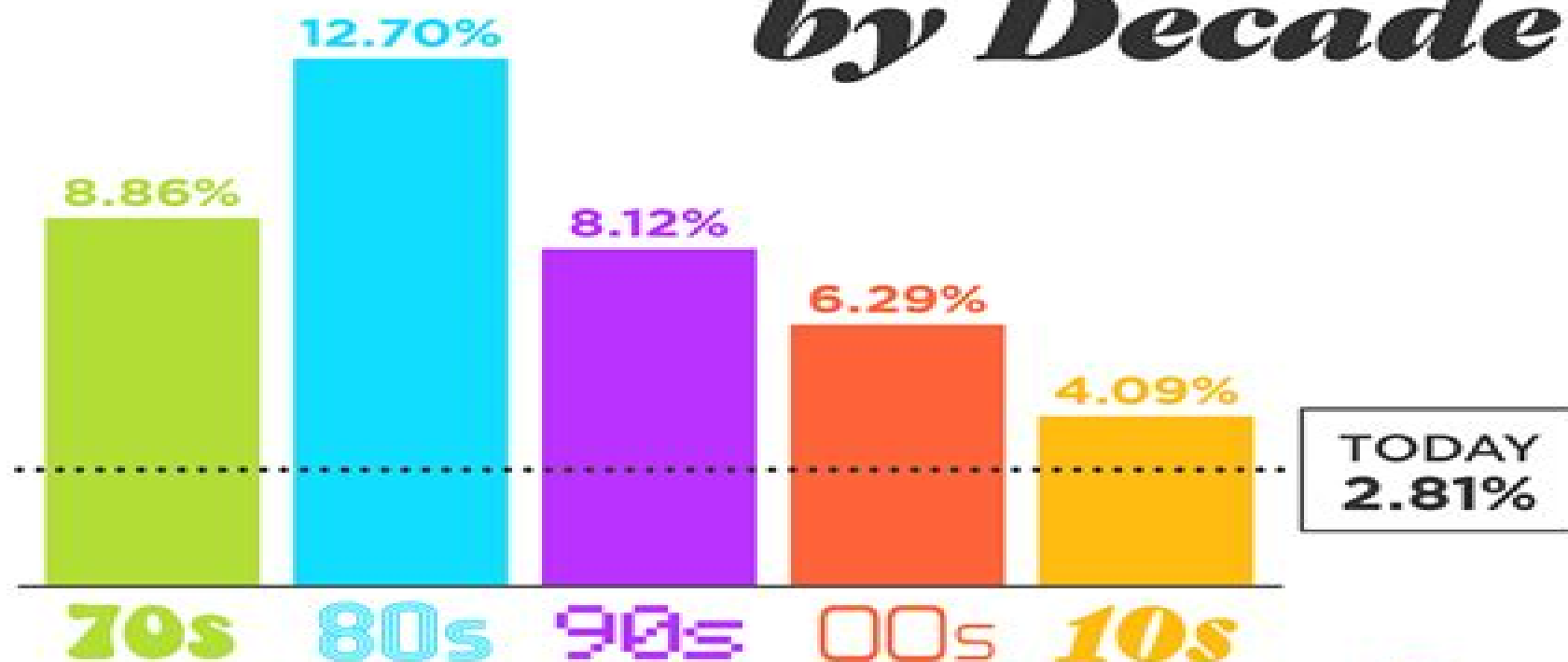


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Source: Freddie Mac



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Thomas Griffiths

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the overall reaction is exothermic

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web a2 1 st b2 1 st kinetics practice problems and solutions d write the rate law for the overall reaction rate $k_a = 2.9$ consider the following mechanism $\text{o}_3 \rightarrow \text{o} + \text{o}_2$ fast $\text{o} + \text{o}_3 \rightarrow 2 \text{o}_2$ slow a write the overall balanced chemical equation $2 \text{o}_3 \rightarrow 3 \text{o}_2$ b identify any intermediates within the mechanism c

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web reaction rates google classroom $\text{mg} + 2 \text{hcl} \rightarrow \text{mgcl}_2 + \text{h}_2$ the reaction between mg and hcl is represented by the equation above in a kinetics experiment a 0.080 g sample of mg ribbon is added to an erlenmeyer flask containing 50 ml of 3.0 M hcl at 25 °C

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