

SAT PRACTICE REVIEW & VOCAB NOTES

Name: _____ | Last Updated: _____

SAT PRACTICE TEST - [Test Number or Date]

☰ Key Questions & Answers

• Reading

- Q15: "According to the passage, the author's attitude toward..."
→ Correct Answer: B - Based on lines 22-25, which show a sarcastic tone.
- Q22: "What does the word 'dissent' most nearly mean in line 34?"
→ Correct Answer: D - "Dissent" - disagreement.

• Math

- Q12: System of equations involving variable m , eliminate x
→ Correct Answer: C - Subtracting the equations gives the result.

☰ Vocabulary Notes

Word	Meaning (English)	Memory Tip
dissent	disagreement or opposition	Think: "disagree"
ubiquitous	present everywhere	e.g., "WiFi is ubiquitous"

Additional Notes

- Reading section included many academic words - practice more literary passages.
- Math: Struggled with square root expressions - review radicals and simplification.

Sat Practice Review Setup

DJ Losen



Sat Practice Review Setup:

Unveiling the Energy of Verbal Artistry: An Psychological Sojourn through **Sat Practice Review Setup**

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 $4\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + 1\text{PbCl}_2$
 $6\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 4\text{KBr} + 1\text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $4\text{CH}_4 + 2\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$
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 $1\text{O}_2 + 3\text{H}_2 \rightarrow 3\text{H}_2\text{O}$
 $2\text{O}_2 + 3\text{H}_2 \rightarrow 2\text{H}_2\text{O} + 4\text{H}_2$
 $\text{rbh} + 2\text{H}_2\text{O} \rightarrow \text{rboh} + 2\text{H}_2$
 $5\text{Cl}_2 + 2\text{KBr} \rightarrow 2\text{KCl} + \text{Br}_2$
 $6\text{N}_2 + 2\text{H}_2 \rightarrow 3\text{N}_2 + 7\text{Ndf}$
 $3\text{Ca} + \text{Nd} \rightarrow \text{Caf} + 2\text{H}_2$
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 $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $\text{Pb} + \text{HCl} \rightarrow 2\text{H}_2 + 1\text{PbCl}_2$
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 $1\text{C}_4\text{H}_{10} + 6\text{O}_2 \rightarrow 1\text{H}_2\text{O} + 2\text{C}_2\text{H}_4\text{O}_2$

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 $2\text{NH}_3 + 3\text{H}_2 + \text{O}_2 \rightarrow 4\text{Al} + 3\text{H}_2\text{SO}_4 + 2\text{Al}_2\text{SO}_4 + 3\text{H}_2$
 $42\text{P}_4 + 10\text{O}_2 \rightarrow 3\text{H}_2\text{PO}_3 + 43\text{H}_2\text{O} + 3\text{O}_2 + 4\text{HNO}_2$

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