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R Bogdan



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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 $\text{Fe} + \text{Co} \rightarrow \text{Fe}^{2+} + \text{Co}^{2+}$ oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ag} + \text{Ni}^{2+}$ oxidation reduction 25 $\text{Cu} + \text{Pb} \rightarrow \text{Cu}^{2+} + \text{Pb}$ oxidation reduction 26 O_2

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