Annotated periodic table of the elements

Group names*

1 Hydrogen & the alkali metals
2 Alkaline earth metals
11 Coinage metals
12 Volatile metals
13 Icosagens
14 Crystallogens
15 Pnictogens
16 Chalcogens
17 Halogens
18 Helium group

*Groups 3 to 10 are named after their first members i.e. Group 3 is the Scandium Group

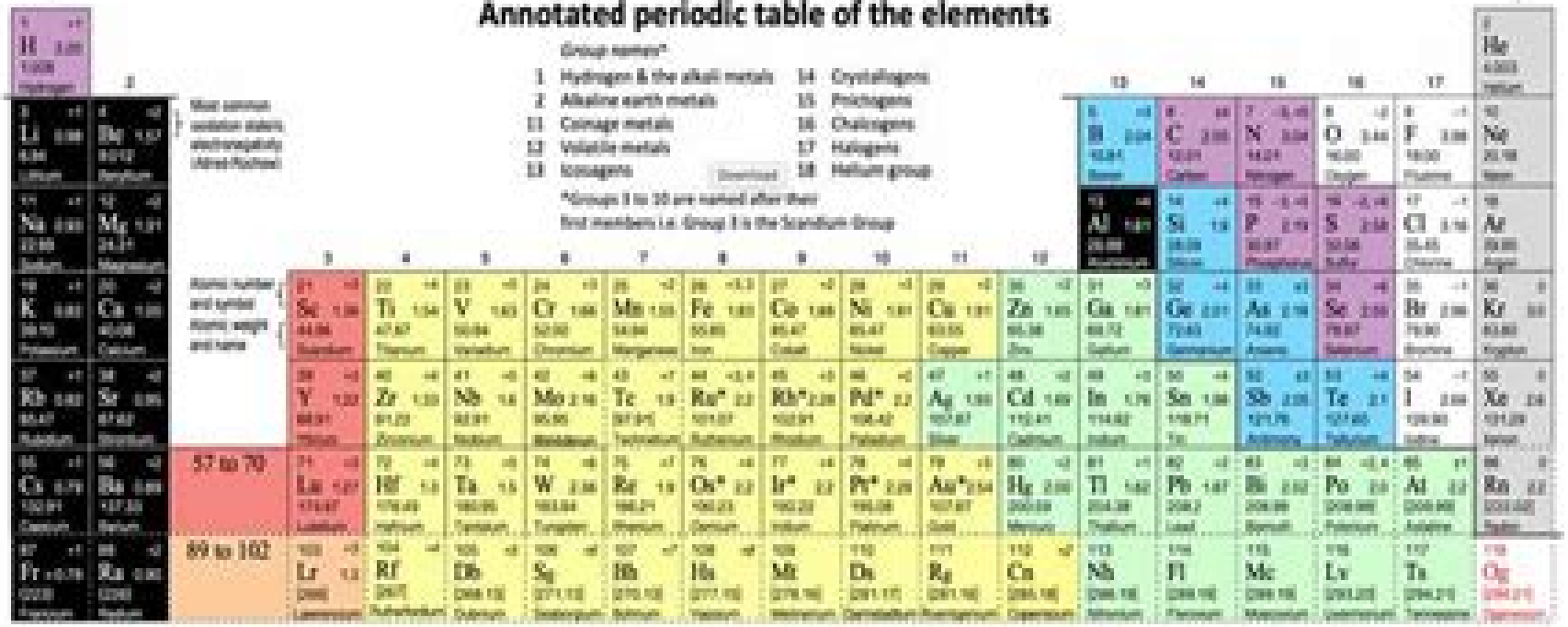
Download

Most common oxidation states, electronegativity (Allred-Rochow)

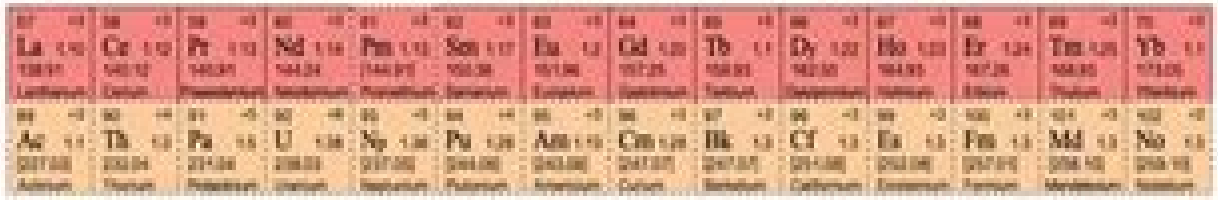
Atomic number and symbol, Atomic weight and name

57 to 70

89 to 102



CLASS COUNTERPARTS		
Metals		Nonmetals
Active	Pre-transition	Corrosive
	Rare earth	
	Actinide	
Transition		Intermediate
Frontier		Metalloid
Noble (Ru-Pt; Os-Au)		Noble gas



Radioactive

Oganesson is likely to be either a metalloid or a post-transition metal

Metals | Nonmetals

Active metals

Pre-transition metals
Light, reactive, mostly of low mechanical strength, melting and boiling points. The group 1 members are easily cut with a knife. Only Be, Mg and Al have any structural uses. Most form basic oxides (Be and Al are amphoteric).

Rare earths
Closely resemble Ca, Sr and Ba, but many are heavier, and they form mostly pale-coloured compounds. Similar to one another, esp. La to Lu (aka the lanthanides), and hard to separate for that reason. Abundant but widely dispersed; commercially viable concentrations are rare. Oxides are strongly to moderately basic. Sc, Y& and Lu exhibit weak amphoterism.

Actinides
Soft, dense and reactive metals, those from Es onwards are synthetic. Early actinides (Th-Am) show some similarities to transition metals and have basic, amphoteric, or acidic oxides; late actinides are more like lanthanides. Many actinides form coloured compounds.

Transition metals
Moderate to high density metals, with high melting and boiling points; many have high hardness and mechanical strength. Those from Rf onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.

Noble metals
Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalytic properties. Very weakly electropositive; insoluble oxides.

Frontier metals
Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater complexity or fewer nearest neighbours than other metals. All used or trialed as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterised, to varying degrees, by covalent bonding tendencies, amphoterism, and anionic species such as zincates, stannites, and bismuthates.

Metalloids
Metallic looking, brittle, fair electrical conductors. All known in semiconducting forms. Moderate ionisation energies (750–1,000 kJ/mol), and electronegativity values (1.9–2.0). Mostly behave chemically as non-metals. Can form alloys with metals. Amphoteric or weakly acidic oxides.

Intermediate nonmetals
H and N are colorless gases; C, P*, and Se are metallic looking solids while S is yellow. The latter four are brittle but all known in malleable, pliable or ductile forms. Moderate to high ionisation energies, low to high electron affinities, moderate to high electronegativity values, relatively poor to moderately strong oxidising agents, tend to form covalent or metallic to near-metallic compounds with metals. Their oxides are acidic, or neutral (H_2O, CO, NO, N_2O).

* as black P, the most stable form in ambient conditions

Group 3
This starts as Sc-Y, and then bifurcates into a La branch i.e. La-Ac, and a Lu branch i.e. Lu-Lr. Trends going down Sc-Y and the

- La branch resemble those of groups 1–2
- Lu branch more resemble those of groups 8–10

Corrosive nonmetals
O, F, and Cl are, respectively, colorless, pale yellow, and pale yellow-green gases; Br is red-brown liquid; I is a metallic looking solid. Chemically tend to have to high ionisation energies, electron affinities, and electronegativity values, and be relatively strong oxidising agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically corrosive nature of the non-metallic halogens. All five tend to form predominately ionic compounds with metals. Their oxides are acidic.

Noble gases
Colorless, odorless, non-flammable gases with very low chemically reactivity. The first noble gas compound, of approximate composition $XePtF_6$, was not prepared until 1962. More than half a thousand noble gas compounds were synthesised in the ensuing half-century. Compounds of He, Ne, and Ar, in ambient conditions, are not known.

February 4th, 2020

Periodic Table Unscrambled Table Answers

Jan Rader,Jan Ray

Periodic Table Unscrambled Table Answers:

Microsoft Word Simple Projects Jan Rader,Jan Ray,2001-04 Projects for language arts social studies science and math Provided templates can be modified to meet specific needs Project samples also provided **Periodic Table** Cindy Blobaum,2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket *Scientifica* ,2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities **Unscramble Your Nest Egg** John J. Cunningham,2013-10-25 Pay off mortgage by age 50 55 deposit former mortgage payments into income earning accounts and result will be a cash nest egg big enough to pay 50 percent of living expenses during 20 years of retirement **New Modular Science for GCSE.** Ann Fullick,1998 This homework book provides graded homework questions to help the teacher set homework for a range of abilities It offers 14 sets of homework questions for each of the 12 modules in the NEAB science syllabus **Scientifica Pupil Book 8 (Levels 4-7)** David Sang,Lawrie Ryan,Jane Taylor,Peter Ellis,2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities *The Science Teacher* ,2008 **Cyber Science 6 Tm' 2007 Ed.** , **Pathways in Science: Chemistry (Pt. 1-3)** Joseph M. Oxenhorn,1970 Verbal Reactions - Word Scrambles With a Chemical Flavor (Medium) Chris McMullen,Carolyn Kivett,2011-12-09 VErBAl ReAcTiONS are a new form of word scrambles with a chemical flavor You don t need to know any science to be able to solve VErBAl ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions EXAMPLE Es 2 S P Si 2 O N __ __ __ __ __ __ __ __ This VErBAl ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that s why there are 8 blanks in the puzzle ANSWER For this puzzle the answer is P O S S Es Si O N possession These VErBAl ReAcTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VErBAl ReAcTiON by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using symbols from the

periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Medium volume consists of words with 6 symbols which involves familiarity with common 6 to 12 letter words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U __ __ __ __ 2 2 C N 2 I P __ __ __ __ __ __ 3 Ti C Cr P Y __ __ __ __ __ 4 2 C U 2 S Es __ __ __ __ __ __ You can find the answers below Note that this medium volume consists of chemical words with 6 symbols There are also easy and hard puzzles available in separate volumes which consist of shorter and longer chemical words ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS *Story of the World Activity Book 4 Modern Age* Susan Wise Bauer,2005-11-29 Presents a history of the ancient world from 6000 B C to 400 A D *Verbal Reactions - Word Scrambles With a Chemical Flavor (Hard)* Chris Mcmullen,Carolyn Kivett,2011-06-18 VErBAl ReAcTiONS are a new form of word scrambles with a chemical flavor You don t need to know any science to be able to solve VErBAl ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions Here is a sample Es 2 S P Si 2 O N __ __ __ __ __ __ __ __ This VErBAl ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that s why there are 8 blanks in the puzzle For this puzzle the answer is P O S S Es Si O N possession These VErBAl ReAcTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VErBAl ReAcTiON by rearranging the elements and writing them on the blanks on the right side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using symbols from the periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Hard volume consists of words with 7 to 8 symbols which involves familiarity with common 8 to 14 letter words Other Medium and Easy volumes consist of shorter words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U __ __ __ __ 2 2 C N 2 I P __ __ __ __ __ __ 3 Ti C Cr P Y __ __ __ __ __ 4 2 C U 2 S Es __ __ __ __ __ __ You can find the answers at the end of this paragraph Note that this hard volume consists of chemical words with 7 to 8 symbols which are longer than the examples shown here We recommend starting with our easy or medium puzzles before tackling these hard puzzles available in separate volumes ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS *Jumpstarters for Properties*

of Matter, Grades 4 - 12 Eric T. Olson,2008-12-19 Engage students in describing and comparing general properties of objects like size shape and temperature Then move on to specific types of matter that have characteristic properties such as density and the melting point And lastly have students focus on the notions of elements and compounds Each reproducible page includes five exercises that can be used as whole worksheets for homework assignments and extra practice or cut apart for daily warm up activities Supports NSE standards **Amazing People: Scientists and Mathematicians** Elise Craver,Malaski,Malay,Hailey Scragg,2023-01-23 Inspirational 1st 2nd and 3rd grade workbooks for kids ages 7 Motivate and inspire children with the Amazing People Scientists and Mathematicians Activity Book 1st 2nd and 3rd grade books are a great way for children to learn about inspirational people who have made important contributions to science and math through a variety of activities that are both fun AND educational Why You ll Love This 1st Grade Third Grade Workbook Engaging and educational activities and wall d cor The activity book features biographies activities and flash cards covering 53 inspirational mathematicians and scientists The activity book also includes 1 inspirational poster great for both homeschool and classroom curriculum and classroom d cor Tracking progress along the way Test your child s knowledge with the flash cards that cover scientists and mathematicians learned throughout the book Use the stickers to reward students on a job well done after completing activities Practically sized for every activity The 256 page 1st grade workbook is sized at 7 75 x 10 625 giving your child plenty of space to complete each exercise About Carson Dellosa For more than 45 years Carson Dellosa has provided solutions for parents and teachers to help their children get ahead and exceed learning goals Carson Dellosa supports your child s educational journey every step of the way The Grade 1 Grade 3 Workbook Contains Biographies and activities covering 53 people 64 flash cards 53 stickers 1 motivational poster *KIRIBATI* NARAYAN CHANGDER,2023-01-13 IF YOU ARE LOOKING FOR A FREE PDF PRACTICE SET OF THIS BOOK FOR YOUR STUDY PURPOSES FEEL FREE TO CONTACT ME cbsenet4u gmail com I WILL SEND YOU PDF COPY THE KIRIBATI MCQ MULTIPLE CHOICE QUESTIONS SERVES AS A VALUABLE RESOURCE FOR INDIVIDUALS AIMING TO DEEPEN THEIR UNDERSTANDING OF VARIOUS COMPETITIVE EXAMS CLASS TESTS QUIZ COMPETITIONS AND SIMILAR ASSESSMENTS WITH ITS EXTENSIVE COLLECTION OF MCQS THIS BOOK EMPOWERS YOU TO ASSESS YOUR GRASP OF THE SUBJECT MATTER AND YOUR PROFICIENCY LEVEL BY ENGAGING WITH THESE MULTIPLE CHOICE QUESTIONS YOU CAN IMPROVE YOUR KNOWLEDGE OF THE SUBJECT IDENTIFY AREAS FOR IMPROVEMENT AND LAY A SOLID FOUNDATION DIVE INTO THE KIRIBATI MCQ TO EXPAND YOUR KIRIBATI KNOWLEDGE AND EXCEL IN QUIZ COMPETITIONS ACADEMIC STUDIES OR PROFESSIONAL ENDEAVORS THE ANSWERS TO THE QUESTIONS ARE PROVIDED AT THE END OF EACH PAGE MAKING IT EASY FOR PARTICIPANTS TO VERIFY THEIR ANSWERS AND PREPARE EFFECTIVELY *Standard Catalog for High School Libraries* H.W. Wilson Company,1993 Each vol is divided into 2 parts 1st 7th ed Dictionary catalog and Classified catalog 8th 9th ed have 3rd part Directory of publishers Science

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