

Chemical Reactions: The Nine Bottle Problem

14. The following table summarizes the laboratory experiment. The first question concerns the effect of the concentration of the reactants on the rate of the reaction. The second question concerns the effect of the concentration of the reactants on the equilibrium constant.
- | Run | Initial [A] (M) | Initial [B] (M) | Initial [C] (M) | Initial [D] (M) | Initial [E] (M) | Initial [F] (M) | Initial [G] (M) | Initial [H] (M) | Initial [I] (M) | Initial [J] (M) | Initial [K] (M) | Initial [L] (M) | Initial [M] (M) | Initial [N] (M) | Initial [O] (M) | Initial [P] (M) | Initial [Q] (M) | Initial [R] (M) | Initial [S] (M) | Initial [T] (M) | Initial [U] (M) | Initial [V] (M) | Initial [W] (M) | Initial [X] (M) | Initial [Y] (M) | Initial [Z] (M) | |
|-----|-----------------|------|
| 1 | 0.10 | 0.10 |
| 2 | 0.20 | 0.20 |
| 3 | 0.30 | 0.30 |
| 4 | 0.40 | 0.40 |
| 5 | 0.50 | 0.50 |
| 6 | 0.60 | 0.60 |
| 7 | 0.70 | 0.70 |
| 8 | 0.80 | 0.80 |
| 9 | 0.90 | 0.90 |
| 10 | 1.00 | 1.00 |
- Graphs of the products vs. time are shown for the first question. Graphs of the products vs. time are shown for the second question. Graphs of the products vs. time are shown for the third question. Graphs of the products vs. time are shown for the fourth question. Graphs of the products vs. time are shown for the fifth question. Graphs of the products vs. time are shown for the sixth question. Graphs of the products vs. time are shown for the seventh question. Graphs of the products vs. time are shown for the eighth question. Graphs of the products vs. time are shown for the ninth question. Graphs of the products vs. time are shown for the tenth question.

Source: <http://www.census.gov/hhes/education/data/tables/2000/2000.ed.education.attainment.attainment.html>

Nine Bottle Problem Answers

VM Jensen



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