

Periodic Table of Elements

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18				
1	1 H Hydrogen (1.008)	Atomic # Symbol Name State C Solid Hg Liquid H Gas Rf Unknown Alkali metals Alkaline earth metals Lanthanoids Actinoids Transition metals Poor metals Other nonmetals Noble gases																	3 He Helium (4.003)			
2	2 Li Lithium (6.941)	4 Be Beryllium (9.012)															5 B Boron (10.81)	6 C Carbon (12.011)	7 N Nitrogen (14.007)	8 O Oxygen (15.999)	9 F Fluorine (18.998)	10 Ne Neon (20.180)
3	11 Na Sodium (22.990)	12 Mg Magnesium (24.305)															13 Al Aluminum (26.982)	14 Si Silicon (28.086)	15 P Phosphorus (30.974)	16 S Sulfur (32.06)	17 Cl Chlorine (35.45)	18 Ar Argon (39.948)
4	19 K Potassium (39.098)	20 Ca Calcium (40.078)	21 Sc Scandium (44.956)	22 Ti Titanium (47.88)	23 V Vanadium (50.942)	24 Cr Chromium (51.996)	25 Mn Manganese (54.938)	26 Fe Iron (55.845)	27 Co Cobalt (58.933)	28 Ni Nickel (58.693)	29 Cu Copper (63.546)	30 Zn Zinc (65.38)	31 Ga Gallium (69.723)	32 Ge Germanium (72.64)	33 As Arsenic (74.922)	34 Se Selenium (78.96)	35 Br Bromine (79.904)	36 Kr Krypton (83.80)				
5	37 Rb Rubidium (85.468)	38 Sr Strontium (87.62)	39 Y Yttrium (88.906)	40 Zr Zirconium (91.224)	41 Nb Niobium (92.906)	42 Mo Molybdenum (95.94)	43 Tc Technetium (98.906)	44 Ru Ruthenium (101.07)	45 Rh Rhodium (102.91)	46 Pd Palladium (106.36)	47 Ag Silver (107.87)	48 Cd Cadmium (112.41)	49 In Indium (114.82)	50 Sn Tin (118.71)	51 Sb Antimony (121.76)	52 Te Tellurium (127.6)	53 I Iodine (126.91)	54 Xe Xenon (131.29)				
6	55 Cs Cesium (132.91)	56 Ba Barium (137.33)	57-71		72 Hf Hafnium (178.49)	73 Ta Tantalum (180.95)	74 W Tungsten (183.84)	75 Re Rhenium (186.21)	76 Os Osmium (190.23)	77 Ir Iridium (192.22)	78 Pt Platinum (195.08)	79 Au Gold (196.97)	80 Hg Mercury (200.59)	81 Tl Thallium (204.38)	82 Pb Lead (207.2)	83 Bi Bismuth (208.98)	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)			
7	87 Fr Francium (223)	88 Ra Radium (226)	89-103		104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (266)	107 Bh Bohrium (264)	108 Hs Hassium (277)	109 Mt Meitnerium (268)	110 Ds Darmstadtium (271)	111 Rg Roentgenium (272)										

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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La Lanthanum (138.91)	Ce Cerium (140.12)	Pr Praseodymium (140.91)	Nd Neodymium (144.24)	Pm Promethium (145)	Sm Samarium (150.36)	Eu Europium (151.96)	Gd Gadolinium (157.25)	Tb Terbium (158.93)	Dy Dysprosium (162.50)	Ho Holmium (164.93)	Er Erbium (167.26)	Tm Thulium (168.93)	Yb Ytterbium (173.05)	Lu Lutetium (174.97)
Ac Actinium (227)	Th Thorium (232.04)	Pa Protactinium (231.04)	U Uranium (238.03)	Np Neptunium (237)	Pu Plutonium (244)	Am Americium (243)	Cm Curium (247)	Bk Berkelium (247)	Cf Californium (251)	Es Einsteinium (252)	Fm Fermium (257)	Md Mendelevium (258)	No Nobelium (259)	Lr Lawrencium (262)

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Conceptual Chemistry Volume I For Class XI S.K. Jain & Shailesh K. Jain,1998 Conceptual Chemistry Volume I For Class XI **Educart CBSE Class 11 Chemistry Question Bank 2026 (Strictly for 2025-26 Exam)** Educart,2025-06-07 The Educart CBSE Class 11 Chemistry Question Bank 2026 is specially designed for students preparing for the 2025 26 session This book follows the latest CBSE syllabus and exam guidelines to help students build strong concepts and prepare well for their school exams Key Features 100% Based on Latest CBSE Syllabus Strictly follows the official CBSE Class 11 Chemistry syllabus for the 2025 26 academic year Chapterwise and Topicwise Questions Covers all chapters with a variety of CBSE type questions MCQs Very Short Short and Long Answer Assertion Reason and Case Based questions NCERT Focused Practice All questions are based on the NCERT Class 11 Chemistry textbook ensuring no confusion during school assessments Fully Solved Answers Includes complete step by step CBSE marking scheme solutions for all questions to help students learn how to write accurate answers in exams Competency Based Questions Questions framed to build understanding of real life applications and concepts as recommended by the new CBSE paper pattern Self Evaluation Tools Includes chapter tests and sample practice questions for every chapter to test preparation This book is a complete practice resource for Class 11 Chemistry students It is suitable for classwork homework and revision before school tests and final exams If you re looking for a reliable exam focused question bank to help you study smarter the Educart Class 11 Chemistry Question Bank is a smart choice *Foundation Course in Chemistry for JEE/ NEET/ Olympiad Class 10 with Case Study Approach - 5th Edition* Disha Experts,2020-04-06 Foundation Course in Chemistry for JEE NEET Olympiad Class 10 with Case Study Approach is the thoroughly revised and updated 5th edition 2 colour of the comprehensive book for Class 10 students who aspire to become Doctors Engineers The book is focused at 3 Goals Bring Concept Clarity Sharpen Problem Solving Build a Strong Foundation The book discusses theoretical concepts in detail accompanied by Illustrations Learn More Let s Do Activity Did You Know Time to Check your Knowledge Another unique feature of this book is the Case Study Approach where most critical Problem Solving Concepts are discussed in various Permutations and Combinations so as improve Problem Solving Skills among the students The theory is followed by the Exercise part which covers in total 1800 questions divided into 4 levels of fully solved exercises which are graded as per their level of difficulty Exercise 1 Master Boards MCQs FIB True False Assertion Reason Passage Matching Very Short Short Long Answer Type Questions including Past Years Board Qns This Exercise also includes Reasoning Based HOTS and Case Based MCQs Exercise 2 Master the

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The Mathematics of the Periodic Table D. H. Rouvray, R. Bruce King, 2006 The Periodic Table effectively embraces the whole realm of chemistry within the confines of one comparatively simple and easily understood chart of the chemical elements Over many years the Periodic Table has proven to be indispensable not only to chemists of all kinds but also to a host of other scientists including biologists geologists and physicists It is thus hardly surprising that the Periodic Table has become one of our most celebrated contemporary scientific icons In the present work various aspects of the Periodic Table that are seldom if ever featured elsewhere are given prominence The twelve presentations contained herein all have a mathematical flavour because it is the intention to highlight the often neglected mathematical features of the Periodic Table and several closely related topics The book starts out by considering predictions of what the ultimate size of the Periodic Table will be when all of the possible artificial chemical elements have been synthesised It then moves on to an examination of the nature of the periodicity extant in the Periodic Table and some methods for the prediction of the properties of the super heavy elements The Periodic Table is next explored in various dimensions other than two The natural clustering of the elements into groups is studied by three different but complementary routes namely via the topological structures of the groups the self association of the elements as evidenced by neural network studies and information theoretical analysis of the behaviour of atoms Following a detailed investigation of the mathematical basis for the periodicity seen in atomic and molecular spectroscopy three separate presentations delve into many different aspects of the group theoretical structure of the Periodic Table The unusual combination of themes offered here will appeal to all who seek a more detailed and intimate knowledge of the Periodic Table than that available in standard texts on the subject

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The Encyclopaedia Britannica James Louis Garvin, Franklin Henry Hooper, Warren E. Cox, 1929 *Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed* ,2000 **Encyclopaedia Britannica** ,1929 **New Jersey Science Curriculum Framework** ,1998 **Scientific American** ,1910 **A Class-book of Chemistry ...** G. C. Donington, 1920 Oswaal NCERT Exemplar (Problems - Solutions) Class 11 Physics, Chemistry and Biology (Set of 3 Books) For 2024 Exam Oswaal Editorial Board, 2023-10-28 Description of the product Chapter wise and Topic wise presentation Chapter wise Objectives A sneak peek into the chapter Mind Map A single page snapshot of the entire chapter Revision Notes Concept based study materials Tips Tricks Useful guidelines for attempting each question perfectly Some Commonly Made Errors Most common and unidentified errors are focused Expert Advice Oswaal Expert Advice on how to score more Oswaal QR Codes For Quick Revision on your Mobile Phones and Tablets **Scientific American Supplement** ,1919 **Transactions of the Illinois State Academy of Science** Illinois State Academy of Science, 1964 **School Science and Mathematics** ,1927

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