

Personal Biography Template

Christine Duncan grew up as an only child. Since both her parents didn't have any siblings either, Christine also grew up as an only grandchild. She grew up in a small neighborhood with adults—her parents and grandparents.

Because of this, she never really knew what it meant to be a kid. Since the adults around Christine were quite overprotective too, she didn't learn how to relate well with other kids.

Growing up, Christine felt accustomed to the company of adults. She was always more mature than the other children at school. Even when she gained friends, she often felt annoyed at them while growing up because she felt that they were too childish.

The adults who raised Christine had invested a lot in her—this encouraged her to excel in all of her subjects as well as any hobbies she decided to take up. Now that she's all grown up, accepts that back then, she may have seemed smug to her peers.

While she didn't realize this back then, she hadn't learned self-awareness until years later. When Christine was in high school, she started feeling the pressure from her parents' and grandparents' high expectations. Gradually, she stopped joining extra-curricular activities and instead, made the choice to work.

Still, Christine graduated at the top of her class, although she wasn't the valedictorian. Looking back, Christine thinks that she didn't try as hard because she didn't want to give her family the satisfaction.

Then when it was time for college, she chose to go to a public university that was far from her home instead of the private university her parents hoped she would attend. She made this choice so she could feel like a "small fish in a big pond."

Throughout her college life, Christine didn't join a lot of outside activities. She joined a sorority and even became the president of the group after several of her sisters convinced her to take the role.

She also worked to pay her bills and, after several jobs, she started working for an investment firm. From there, she aimed to meet her potential, take on new challenges, but she didn't strive hard to go beyond expectations.

Probably because of the kind of family she grew up in, Christine never imagined herself having a family of her own. She didn't want to get tied down by a husband and children, so she always believed that marriage wasn't for her.

But in the year 2010, she met a man named Don and—after some happy years together—she decided to tie the knot with him. One year later, they welcomed their first daughter into their life. Since then, she let go of her career-woman life and became a stay-at-home mom. Christine feels both happy and proud of this role of hers.

She feels like this is the perfect role for her as she watches her young daughter grow into a happy little girl. Christine remains driven but this time, not for herself, but for her daughter. Although she still finds it challenging to relate to children, she feels like she is an expert when it comes to her child.

Personal Biography Template For Students

Liying Dong



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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- \text{aq} + 2\text{Ag}^+ \text{aq}$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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