

Key.

b) 6.45 moles Cr_2O_3 $\text{FW} = 152.0 \text{ g/mol}$ 980 g

$$6.45 \text{ mol} \times \left(\frac{152.0 \text{ g}}{1 \text{ mol}} \right) = 980 \text{ g}$$

c) 0.144 mole K_2S $\text{FW} = 110.3 \text{ g/mol}$ 15.9 g

$$0.144 \text{ mol} \times \left(\frac{110.3 \text{ g}}{1 \text{ mol}} \right) = 15.9 \text{ g}$$

d) 5.0 moles $\text{C}_6\text{H}_{12}\text{O}_6$ 180.0 g/mol 900 g $\approx 9.0 \times 10^2 \text{ g}$

$$5.0 \text{ mol} \times \left(\frac{180.0 \text{ g}}{1 \text{ mol}} \right) = 900 \text{ g}$$

e) 5.10×10^{-4} mole NaCl $\text{FW} = 58.5 \text{ g/mol}$ $0.0298 \text{ g} \approx 2.98 \times 10^{-2} \text{ g}$

$$5.10 \times 10^{-4} \text{ mol} \times \left(\frac{58.5 \text{ g}}{1 \text{ mol}} \right) = 0.0298 \text{ g}$$

Solve the following stoichiometry problems. Write a balanced equation for each.

a) How many moles of hydrogen gas are needed to react with 15.1 g chlorine gas, to produce hydrogen chloride gas? Equation: $\text{H}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightarrow 2 \text{HCl}(\text{g})$

$$15.1 \text{ g} \times \left(\frac{1 \text{ mol}}{71.0 \text{ g}} \right) = 0.213 \text{ mol } \text{Cl}_2$$

$$0.213 \text{ mol } \text{Cl}_2 \left(\frac{1 \text{ mol H}_2}{1 \text{ mol Cl}_2} \right) = 0.213 \text{ mol H}_2 \quad \underline{0.213 \text{ mol H}_2}$$

b) How many grams of oxygen gas are needed to react completely with 16.2 g of hydrogen gas, to produce water? Equation: $2 \text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2 \text{H}_2\text{O}(\text{l})$

$$16.2 \text{ g} \times \left(\frac{1 \text{ mol}}{2.0 \text{ g}} \right) = 8.10 \text{ mol H}_2$$

$$8.10 \text{ mol H}_2 \left(\frac{1 \text{ mol O}_2}{2 \text{ mol H}_2} \right) = 4.05 \text{ mol O}_2 \quad \underline{130 \text{ g O}_2}$$

$$4.05 \text{ mol O}_2 \left(\frac{32.0 \text{ g/mol}}{1 \text{ mol}} \right) = 130 \text{ g O}_2$$

c) How many moles of calcium oxide are produced when 36.5 g of calcium react completely with oxygen gas? Equation: $2 \text{Ca}(\text{s}) + \text{O}_2(\text{g}) \rightarrow 2 \text{CaO}(\text{s})$

$$36.5 \text{ g} \left(\frac{1 \text{ mol}}{40.1 \text{ g}} \right) = 0.910 \text{ mol Ca}$$

$$0.910 \text{ mol Ca} \left(\frac{2 \text{ mol CaO}}{2 \text{ mol Ca}} \right) = 0.910 \text{ mol } \underline{0.910 \text{ mol CaO}}$$

d) How many grams of aluminum are required to produce 410 g of aluminum oxide, through reaction with oxygen gas? Equation: $4 \text{Al} + 3 \text{O}_2 \rightarrow 2 \text{Al}_2\text{O}_3$

$$410 \text{ g} \left(\frac{1 \text{ mol}}{102.0 \text{ g}} \right) = 4.02 \text{ mol Al}_2\text{O}_3$$

$$4.02 \text{ mol Al}_2\text{O}_3 \left(\frac{4 \text{ mol Al}}{2 \text{ mol Al}_2\text{O}_3} \right) = 8.04 \text{ mol Al}$$

$$8.04 \text{ mol Al} \left(\frac{27.0 \text{ g}}{1 \text{ mol}} \right) = 217 \text{ g}$$

$$\underline{217 \text{ g} \approx 220 \text{ g (2sf)}}$$

e) How many grams of solid iodine are produced by the decomposition of 25.4 grams of hydrogen iodide gas? Equation: $2 \text{HI} \rightarrow \text{H}_2 + \text{I}_2$

$$25.4 \text{ g} \left(\frac{1 \text{ mol}}{127.9 \text{ g}} \right) = 0.199 \text{ mol HI}$$

$$0.199 \text{ mol HI} \left(\frac{1 \text{ mol I}_2}{2 \text{ mol HI}} \right) = 0.0993 \text{ mol I}_2 \quad \underline{25.2 \text{ g I}_2}$$

$$0.0993 \text{ mol I}_2 \left(\frac{253.8 \text{ g}}{1 \text{ mol}} \right) = 25.2 \text{ g I}_2$$

Mole Ratio Wkst Answers

Judd E. Hollander

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Mole Ratio Wkst Answers:

Mole Ratio Wkst Answers: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of bestselling books, exploring the captivating narratives that have charmed audiences this year. The Must-Read : Colleen Hoover's "It Ends with Us" This heartfelt tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover masterfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can succeed. Mole Ratio Wkst Answers : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This intriguing historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids compelling storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens crafts a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These popular novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of compelling stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and suspenseful novel that will keep you guessing until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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Table of Contents Mole Ratio Wkst Answers

1. Understanding the eBook Mole Ratio Wkst Answers
 - The Rise of Digital Reading Mole Ratio Wkst Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Mole Ratio Wkst Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mole Ratio Wkst Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mole Ratio Wkst Answers
 - Personalized Recommendations
 - Mole Ratio Wkst Answers User Reviews and Ratings
 - Mole Ratio Wkst Answers and Bestseller Lists
5. Accessing Mole Ratio Wkst Answers Free and Paid eBooks
 - Mole Ratio Wkst Answers Public Domain eBooks
 - Mole Ratio Wkst Answers eBook Subscription Services
 - Mole Ratio Wkst Answers Budget-Friendly Options
6. Navigating Mole Ratio Wkst Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Mole Ratio Wkst Answers Compatibility with Devices
 - Mole Ratio Wkst Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mole Ratio Wkst Answers
 - Highlighting and Note-Taking Mole Ratio Wkst Answers
 - Interactive Elements Mole Ratio Wkst Answers
8. Staying Engaged with Mole Ratio Wkst Answers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mole Ratio Wkst Answers
9. Balancing eBooks and Physical Books Mole Ratio Wkst Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mole Ratio Wkst Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Mole Ratio Wkst Answers
- Setting Reading Goals Mole Ratio Wkst Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Mole Ratio Wkst Answers
- Fact-Checking eBook Content of Mole Ratio Wkst Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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