## Molarity Calculations

Calculate the molarities of the following solutions:

Maci

2.3 moles of sodium chloride in 0.45 liters of solution.

Ca COa

1.2 moles of calcium carbonate in 1.22 liters of solution.

NINSOU

0.09 moles of sodium sulfate in 12 mL of solution.

LiF

0.75 moles of lithium fluoride in 65 mL of solution.

Ma(CH, COO),5)

0.8 moles of magnesium acetate in 5 liters of solution.

(a(NO2), B)

120 grams of calcium nitrite in 240 mL of solution.

NAOH

 $MW_{G(NV_1)_1} = 132.093/met$  N = 0.91 met7) 98 grams of sodium hydroxide 98 grams of sodium hydroxide in 2.2 liters of solution.

1.2 grams of hydrochloric acid in 25 mL of solution.

45 grams of ammonia in 0.75 L of solution.

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# **Molarity Calculations Key Answers Show Work**

William R. Robinson, Jerome D. Odom, Henry Fuller Holtzclaw

## **Molarity Calculations Key Answers Show Work:**

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Studies on Oxidation-reduction I-X National Institutes of Health (U.S.), 1928 Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions Chemistry William R. Robinson, Jerome D. Odom, Henry Fuller Holtzclaw, 1992 to odd numbered problems General Chemistry Kenneth W. Whitten, Raymond E. Davis, 1996 Are you looking for the key to success in your chemistry class In CHEMISTRY you will find a strong molecular reasoning focus problem solving exercises and an innovative online homework management system that will prepare you for any challenge you might encounter The textbook is filled with learning aids Introductory Chemistry Steve Russo, Mike Silver, 2002 Steve Russo and that will help you master concepts of the course Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities Readers make the connections between concepts and the problem solving techniques they need to master as they read The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place tm Special Edition and on The Chemistry of Life CD ROM For college instructors and students

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