



Pumpkin Spice Last 90 Days

Lauren Gardner



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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectives to correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objects can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound

the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

exploring boiling points sharmin khatun chem 1141 studocu - May 17 2023

exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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dec 7 2019 1987 the sample essay on boiling point determination lab report deals with a framework of research based facts approaches and arguments concerning this theme to see the essay's introduction body paragraphs and conclusion read on should be based on observations made during the experiment

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0.5 ml of material

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may 31 2018 in 2012 12 medical schools were opened in ethiopia to tackle the significant shortage of doctors this included aksum school of medicine situated in aksum a rural town in northern ethiopia the new innovative medical curriculum nimc is a four year programme designed by the ethiopian federal ministries of health and education

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a multi center institution based cross sectional study was conducted among 1051 medical graduates selected through cluster sampling method from may-july 2022 in medical schools found in amhara region northwest ethiopia data were collected from

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jan 1 2013 ethiopia has been implementing traditional medical education and graduates were missing some competencies 5 studies elsewhere reported that students were passively engaged in the

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jun 10 2021 registration for 2021 new innovative medical education initiative nimei has already started and deadline is on june 18 2021 so kindly read the requirement criteria for the registration and use link below to register

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results the objectives rationale as well as the approach and methodology used to develop the curriculum are outlined the processes involved in the implementation and the initial appraisal of the developed curriculum was also discussed

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may 31 2018 the new innovative medical curriculum nimc is a four year programme designed by the ethiopian federal ministries of health and education which has been in place in aksum since 2012 there have been 4 years of clinical students on the course with the first cohort of students graduating in 2016

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content has given emphasis to essential clinical and professional skills emergency live saving surgical and obstetrics skills and has introduced ethiopian health problems mainly infectious diseases rural health the main intent of this new approach to medical education has been twofold first to ensure relevance to the task of medical

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sep 5 2019 results 673 99 6 of 676 students approached completed the survey representing 39 5 of all 1705 medical

students enrolled at aau in 2014 most students felt the flooding policy had a negative impact on their medical education and 90 felt there was not adequate infrastructure to support the increased student body

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oct 13 2022 selection of applicants to the new innovative medical education initiative nimei program includes the following criteria the total grade break down is indicated below a written examination 60

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