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option c is incorrect because 566 kj mol is the Δh per mole of o2 and not for 2 moles even though options b and d are expressed in grams they are still correct because 56

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web q 140 g 4 184 j 16oc g oc q 9372 16 j 35 5 g x 1 mole kcn 0 5453 moles 65 1 g 0 5453 mole moles 5 when a 15 g sample of srcl2 dissolves in 80 g of water in a calorimeter the temperature drops from 42 oc to 15 c calculate the Δh for the process thermochemistry answers and explanations integrated mcat - Feb 11 2023

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