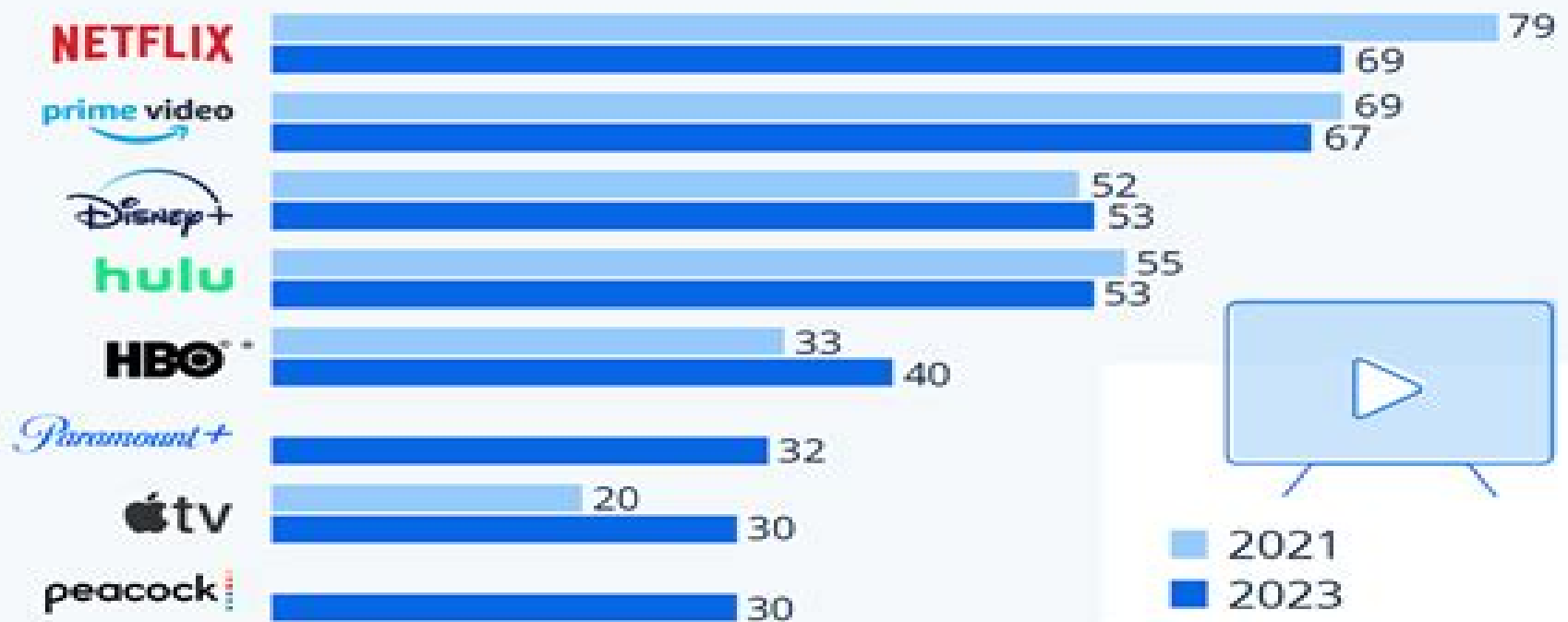


The Most Popular Streaming Services in the U.S.

Share of respondents who in the last 12 months were paying customers at the following streaming providers



* Incl. Max, Go and Now

3,843/6,660 respondents (18-64 y/o) surveyed in the U.S. Jul. 2020/'22-Jun. 2021/'23

Source: Statista Consumer Insights



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Marco Cascella



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Statistics of Power and Machinery Employed in Manufactures United States. Census Office, 1885 **The United States Patents Quarterly** ,1929 Report of cases relating to patents trade marks copyrights decided by Supreme Court of the United States United States Circuit courts of appeals District courts of the United States United States Court of Customs and Patent Appeals Court of Claims of the United States United States Court of Appeals for the District of Columbia Commissioner of Patents and Patent Office Board of Appeals *The Geology of Cowal* William Gunn, Charles Thomas Clough, James Bastian Hill, 1897

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
 one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{NaOH} \cdot \text{Na}_2\text{C}_2\text{O}_4 \cdot 4\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction
 $\text{NaOH} + \text{C}_6\text{H}_4\text{COOH} \rightarrow \text{C}_6\text{H}_4\text{COONa} + \text{H}_2\text{O}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the

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