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ADDITION VERSION 1

$2+4=$ _____

$1+3=$ _____

$5+6=$ _____

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$2+6=$ _____

$8+3=$ _____

$9+3=$ _____

$2+8=$ _____

$7+3=$ _____

$2+1=$ _____

$6+6=$ _____

$9+7=$ _____

$7+7=$ _____

$3+5=$ _____

$5+5=$ _____

$2+3=$ _____

$5+8=$ _____

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Stagram Picks Math Worksheet How To

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 $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 $\text{Pb} + \text{OH}^- + 2\text{HCl} \rightarrow \text{H}_2\text{O} + \text{PbCl}_2$
 $2\text{Al} + 3\text{Br}_2 \rightarrow 2\text{AlBr}_3$
 $3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2\text{SO}_4$
 $3\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
 $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$
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 $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$
 $\text{Mg} + \text{F}_2 \rightarrow \text{MgF}_2$
 $\text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3$
 $\text{LiF} + \text{P}_4\text{O}_{10} \rightarrow 2\text{P}_2\text{O}_5 + 3\text{Li}_2\text{O}$
 $\text{Be} + \text{NO}_3 \rightarrow 2\text{Be} + \text{NO}_2 + \text{H}_2\text{O}$

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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 $2\text{HNO}_3 \rightarrow \text{H}_2\text{O} + 2\text{O}_2 + 3\text{H}_2$
 $2\text{H}_2\text{O} + 2\text{O}_2 + 4\text{H}_2\text{O} \rightarrow \text{H}_2\text{O} + \text{H}_2\text{O} + 5\text{Cl}_2$
 $2\text{KBr} + \text{KCl} \rightarrow 2\text{Br}_2 + 6\text{N}_2$
 $2\text{H}_2 + \text{NH}_3 \rightarrow 7\text{N}_2\text{F}_6$
 $3\text{Ca} + \text{N}_2\text{F}_6 \rightarrow 8\text{CH}_4$
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