

PERIODIC TABLE (BASIC)

Atomic #	Symbol	Atomic Weight	State	Group	Period	Block	Classification
1	H	1.00794	Gas	1	1	s	Nonmetal
2	He	4.002602	Gas	18	1	s	Noble gas
3	Li	6.941	Metal	1	2	s	Alkali metal
4	Be	9.0122	Metal	2	2	s	Alkaline earth metal
5	B	10.811	Metalloid	13	2	p	Metalloid
6	C	12.011	Nonmetal	14	2	p	Nonmetal
7	N	14.007	Gas	15	2	p	Nonmetal
8	O	15.999	Gas	16	2	p	Nonmetal
9	F	18.998	Gas	17	2	p	Nonmetal
10	Ne	20.180	Gas	18	2	p	Noble gas
11	Na	22.990	Metal	1	3	s	Alkali metal
12	Mg	24.305	Metal	2	3	s	Alkaline earth metal
13	Al	26.982	Metal	13	3	p	Post-transition metal
14	Si	28.086	Metalloid	14	3	p	Metalloid
15	P	30.974	Nonmetal	15	3	p	Nonmetal
16	S	32.06	Nonmetal	16	3	p	Nonmetal
17	Cl	35.45	Gas	17	3	p	Nonmetal
18	Ar	39.948	Gas	18	3	p	Noble gas
19	K	39.098	Metal	1	4	s	Alkali metal
20	Ca	40.078	Metal	2	4	s	Alkaline earth metal
21	Sc	44.956	Metal	3	4	d	Transition metal
22	Ti	47.88	Metal	4	4	d	Transition metal
23	V	50.942	Metal	5	4	d	Transition metal
24	Cr	52.00	Metal	6	4	d	Transition metal
25	Mn	54.938	Metal	7	4	d	Transition metal
26	Fe	55.845	Metal	8	4	d	Transition metal
27	Co	58.933	Metal	9	4	d	Transition metal
28	Ni	58.69	Metal	10	4	d	Transition metal
29	Cu	63.546	Metal	11	4	d	Transition metal
30	Zn	65.38	Metal	12	4	d	Transition metal
31	Ga	69.723	Metal	13	4	p	Post-transition metal
32	Ge	72.64	Metalloid	14	4	p	Metalloid
33	As	74.922	Nonmetal	15	4	p	Nonmetal
34	Se	78.96	Nonmetal	16	4	p	Nonmetal
35	Br	79.904	Gas	17	4	p	Nonmetal
36	Kr	83.80	Gas	18	4	p	Noble gas
37	Rb	85.468	Metal	1	5	s	Alkali metal
38	Sr	87.62	Metal	2	5	s	Alkaline earth metal
39	Y	88.906	Metal	3	5	d	Transition metal
40	Zr	91.224	Metal	4	5	d	Transition metal
41	Nb	92.906	Metal	5	5	d	Transition metal
42	Mo	95.94	Metal	6	5	d	Transition metal
43	Tc	98.906	Metal	7	5	d	Transition metal
44	Ru	101.07	Metal	8	5	d	Transition metal
45	Rh	102.91	Metal	9	5	d	Transition metal
46	Pd	106.42	Metal	10	5	d	Transition metal
47	Ag	107.87	Metal	11	5	d	Transition metal
48	Cd	112.415	Metal	12	5	d	Transition metal
49	In	114.82	Metal	13	5	p	Post-transition metal
50	Sn	118.71	Metal	14	5	p	Post-transition metal
51	Sb	121.76	Metalloid	15	5	p	Metalloid
52	Te	127.6	Nonmetal	16	5	p	Nonmetal
53	I	126.905	Nonmetal	17	5	p	Nonmetal
54	Xe	131.29	Gas	18	5	p	Noble gas
55	Ba	137.33	Metal	2	6	s	Alkaline earth metal
56	La	138.905	Metal	3	6	f	Lanthanide
57	Ce	140.12	Metal	4	6	f	Lanthanide
58	Pr	140.908	Metal	5	6	f	Lanthanide
59	Nd	144.24	Metal	6	6	f	Lanthanide
60	Pm	144.913	Metal	7	6	f	Lanthanide
61	Sm	150.36	Metal	8	6	f	Lanthanide
62	Eu	151.964	Metal	9	6	f	Lanthanide
63	Gd	157.25	Metal	10	6	f	Lanthanide
64	Tb	158.925	Metal	11	6	f	Lanthanide
65	Dy	162.50	Metal	12	6	f	Lanthanide
66	Ho	164.930	Metal	13	6	f	Lanthanide
67	Er	167.26	Metal	14	6	f	Lanthanide
68	Tm	168.933	Metal	15	6	f	Lanthanide
69	Yb	173.054	Metal	16	6	f	Lanthanide
70	Lu	174.967	Metal	17	6	f	Lanthanide
71	Hf	178.49	Metal	4	6	d	Transition metal
72	Ta	180.948	Metal	5	6	d	Transition metal
73	W	183.84	Metal	6	6	d	Transition metal
74	Re	186.207	Metal	7	6	d	Transition metal
75	Os	190.23	Metal	8	6	d	Transition metal
76	Ir	192.22	Metal	9	6	d	Transition metal
77	Pt	195.084	Metal	10	6	d	Transition metal
78	Au	196.967	Metal	11	6	d	Transition metal
79	Hg	200.59	Metal	12	6	d	Transition metal
80	Tl	204.38	Metal	13	6	p	Post-transition metal
81	Pb	207.2	Metal	14	6	p	Post-transition metal
82	Bi	208.98	Metal	15	6	p	Post-transition metal
83	Po	209	Metal	16	6	p	Post-transition metal
84	At	210	Metal	17	6	p	Post-transition metal
85	Ac	227	Metal	3	7	f	Actinide
86	Ra	226	Metal	4	7	f	Actinide
87	Fr	223	Metal	5	7	f	Actinide
88	Ra	226	Metal	6	7	f	Actinide
89	Ac	227	Metal	7	7	f	Actinide
90	Th	232.0377	Metal	8	7	f	Actinide
91	Pa	231.03688	Metal	9	7	f	Actinide
92	U	238.02891	Metal	10	7	f	Actinide
93	Np	237.04817	Metal	11	7	f	Actinide
94	Pu	244.0642	Metal	12	7	f	Actinide
95	Am	243.06136	Metal	13	7	f	Actinide
96	Cm	247.07035	Metal	14	7	f	Actinide
97	Bk	247.07035	Metal	15	7	f	Actinide
98	Cf	251.0834	Metal	16	7	f	Actinide
99	Es	252.0834	Metal	17	7	f	Actinide
100	Fm	257.10528	Metal	18	7	f	Actinide
101	Md	258.10528	Metal	19	7	f	Actinide
102	No	259.10528	Metal	20	7	f	Actinide
103	Lr	262.10528	Metal	21	7	f	Actinide

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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