

PERIODIC TABLE (BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
1 H Hydrogen 1.008	2 He Helium 4.0026																	
3 Li Lithium 6.941	4 Be Beryllium 9.0122																	
5 B Boron 10.81	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180													
11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948											
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 52.00	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933	28 Ni Nickel 58.69	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80	
37 Rb Rubidium 85.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.41	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29	
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71 Lanthanides		72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium 209	85 At Astatine 210	
87 Fr Francium 223	88 Ra Radium 226	89-103 Actinides		104 Rf Rutherfordium 261	105 Db Dubnium 262	106 Sg Seaborgium 266	107 Bh Bohrium 264	108 Hs Hassium 277	109 Mt Meitnerium 268	110 Ds Darmstadtium 271	111 Rg Roentgenium 272	112 Cn Copernicium 285	113 Nh Nihonium 286	114 Fl Flerovium 289	115 Uup Ununpentium 288	116 Lv Livermorium 293	117 Ts Tennessine 294	118 Og Oganesson 294

C Solid

Hg Liquid

H Gas

Rf Unknown

Metals

Nonmetals

Alkali metals

Alkaline earth metals

Lanthanides

Actinides

Transition metals

Post-transition metals

Noble gases

Atomic #

Symbol

Name

Weight

1	H	Hydrogen	1.008
2	He	Helium	4.0026
3	Li	Lithium	6.941
4	Be	Beryllium	9.0122
5	B	Boron	10.81
6	C	Carbon	12.011
7	N	Nitrogen	14.007
8	O	Oxygen	15.999
9	F	Fluorine	18.998
10	Ne	Neon	20.180
11	Na	Sodium	22.990
12	Mg	Magnesium	24.305
13	Al	Aluminum	26.982
14	Si	Silicon	28.086
15	P	Phosphorus	30.974
16	S	Sulfur	32.06
17	Cl	Chlorine	35.45
18	Ar	Argon	39.948
19	K	Potassium	39.098
20	Ca	Calcium	40.078
21	Sc	Scandium	44.956
22	Ti	Titanium	47.88
23	V	Vanadium	50.942
24	Cr	Chromium	52.00
25	Mn	Manganese	54.938
26	Fe	Iron	55.845
27	Co	Cobalt	58.933
28	Ni	Nickel	58.69
29	Cu	Copper	63.546
30	Zn	Zinc	65.38
31	Ga	Gallium	69.723
32	Ge	Germanium	72.63
33	As	Arsenic	74.922
34	Se	Selenium	78.96
35	Br	Bromine	79.904
36	Kr	Krypton	83.80
37	Rb	Rubidium	85.468
38	Sr	Strontium	87.62
39	Y	Yttrium	88.906
40	Zr	Zirconium	91.224
41	Nb	Niobium	92.906
42	Mo	Molybdenum	95.94
43	Tc	Technetium	98.906
44	Ru	Ruthenium	101.07
45	Rh	Rhodium	102.91
46	Pd	Palladium	106.42
47	Ag	Silver	107.868
48	Cd	Cadmium	112.41
49	In	Indium	114.818
50	Sn	Tin	118.710
51	Sb	Antimony	121.757
52	Te	Tellurium	127.6
53	I	Iodine	126.905
54	Xe	Xenon	131.29
55	Cs	Cesium	132.905
56	Ba	Barium	137.327
57	La	Lanthanum	138.905
58	Ce	Cerium	140.12
59	Pr	Praseodymium	140.908
60	Nd	Niodymium	144.24
61	Pm	Promethium	144.913
62	Sm	Samarium	150.36
63	Eu	Europium	151.964
64	Gd	Gadolinium	157.25
65	Tb	Terbium	158.925
66	Dy	Dysprosium	162.50
67	Ho	Holmium	164.930
68	Er	Erbium	167.259
69	Tm	Thulium	168.930
70	Yb	Ytterbium	173.054
71	Lu	Lutetium	174.967
72	Hf	Hafnium	178.49
73	Ta	Tantalum	180.948
74	W	Tungsten	183.84
75	Re	Rhenium	186.207
76	Os	Osmium	190.23
77	Ir	Iridium	192.222
78	Pt	Platinum	195.084
79	Au	Gold	196.967
80	Hg	Mercury	200.59
81	Tl	Thallium	204.38
82	Pb	Lead	207.2
83	Bi	Bismuth	208.98
84	Po	Polonium	209
85	At	Astatine	210
86	Rn	Radon	222
87	Fr	Francium	223
88	Ra	Radium	226
89	Ac	Actinium	227
90	Th	Thorium	232.038
91	Pa	Protactinium	231.036
92	U	Uranium	238.029
93	Np	Neptunium	237.048
94	Pu	Plutonium	244.064
95	Am	Americium	243.061
96	Cm	Curium	247.070
97	Bk	Berkelium	247.070
98	Cf	Californium	251.083
99	Es	Einsteinium	252.083
100	Fm	Fermium	257.103
101	Mn	Mendelevium	258.103
102	No	Nobelium	259.108
103	Lr	Lutetium	262.105
104	Rf	Rutherfordium	261
105	Db	Dubnium	262
106	Sg	Seaborgium	266
107	Bh	Bohrium	264
108	Hs	Hassium	277
109	Mt	Meitnerium	268
110	Ds	Darmstadtium	271
111	Rg	Roentgenium	272
112	Cn	Copernicium	285
113	Nh	Nihonium	286
114	Fl	Flerovium	289
115	Uup	Ununpentium	288
116	Lv	Livermorium	293
117	Ts	Tennessine	294
118	Og	Oganesson	294

270

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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Periodic Table (Basic) Speedy Study Guide Speedy Publishing, 2014-07-04 The average chemistry student lives and breathes the information that is depicted on the periodic table. It is the entire comprehensive list of every element we know of and from these elements all things in our known universe evolve. It also lists atomic weight and the amount of electrons each element has in its structure. A pamphlet that depicts this table would be a go-to guide for all chemistry students throughout their schooling and possibly their career. [Periodic Table Basic](#) Speedy Publishing, 2014-09-04

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Speedy Publishing, 2014-12-28 A study guide of chemistry basics would give students bite-sized chunks of information on a complex subject, making the task of learning it easier and more enjoyable. Study guides present detailed information on a topic in a compact and simplistic way, which helps the student grasp more of what they are being taught. Also, the visual aids that usually are found in study guides can be very effective in keeping the attention of the child as well as embedding the material in their memory. [Periodic Table \(Basic\) \(Speedy Study Guides\)](#)

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Study Guide for Introductory Chemistry, a Foundation, Introductory Chemistry, Basic Chemistry Iris Stovall, Steven S. Zumdahl, 1996

TExES Core Subjects 4-8 (211) Book + Online, 2nd Ed. Ann M.L. Cavallo, Karen Allmond, Mary D. Curtis, Marci Smith Deal, Christina Gawlik, Candace Joswick, Melissa Hulings, Kathleen C. Tice, 2023-01-03 REA s TExES Core Subjects 4 8 211 Test Prep with Online Practice Tests Gets You Certified and in the Classroom Teacher candidates seeking certification to teach the middle level grades in Texas s public schools must pass the TExES Core Subjects 4 8 exam Written by a team of faculty experts led by Dr Ann M L Cavallo Associate Dean for Research and Graduate Studies at the University of Texas at Arlington REA s test prep provides extensive coverage of the four core subject areas tested on the exam English Language Arts and Reading 806 Mathematics 807 Social Studies 808 and Science 809 In addition to a thorough review this test prep features a diagnostic test and 2 full length practice test batteries 1 in the book and 1 online at the REA Study Center that deal with every question type competency and skill tested on the exam REA s online tests run under timed conditions and provide automatic scoring and diagnostic feedback on every question to help teacher candidates zero in on the topics that give them trouble now so they can succeed on test day The new 200 question Core Subjects test one of the largest of its kind in the United States was first administered in January 2015 it replaced the

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"O" Level Study Guide - Chemistry Quite Easily Done , Study Guide for Hein and Arena's Foundations of College Chemistry Peter Scott,Rachael Enriquez Porter,1999-04-18

Introduction to Nuclear Engineering Supathorn Phongikaroon,2023-09-29 Introduction to Nuclear Engineering serves as an accompanying study guide for a complete introductory single semester course in nuclear engineering It is structured for general class use alongside fundamental nuclear physics and engineering textbooks and it is equally suited for individual self study The book begins with basic modern physics with atomic and nuclear models It goes on to cover nuclear energetics radioactivity and decays and binary nuclear reactions and basic fusion Exploring basic radiation interactions with matter the book finishes by discussing nuclear reactor physics nuclear fuel cycles and radiation doses and hazard assessment Each chapter highlights basic concepts examples problems with answers and a final assessment The book is intended for first year undergraduate and graduate engineering students taking Nuclear Engineering and Nuclear Energy courses

Teaching Secondary

Science: A Complete Guide Adam Boxer, 2021-11-19 Teaching science is no simple task Science teachers must wrestle with highly abstract and demanding concepts ideas which have taken humanity's greatest minds thousands of years to formulate and refine Communicating these great and awesome theories involves careful forethought and planning We need to deliver crystal clear explanations guide students as they develop their embryonic knowledge and then release them to develop their thinking independently all the while curating and tending to their long term understanding as it develops over time In Teaching Secondary Science A Complete Guide Adam breaks down the complex art of teaching science into its component parts providing a concrete and comprehensive set of evidence informed steps to nurturing brilliant science students Adam hopes that you find this book interesting but his main aim is for you to find it useful Useful when it comes to sketching out your curriculum useful when preparing your explanations useful for mapping out how you will check student understanding and useful for all other aspects of science teaching This is a truly complete guide and science teachers of any experience will find it packed with ideas that are new challenging interesting and most importantly useful [Student Study Guide to accompany Foundations of College Chemistry, 11th Edition](#) Morris Hein, Susan Arena, 2003-09-03 Work more effectively and gauge your progress along the way This Student Study Guide that accompanies Hein's Foundations of College Chemistry 11th Edition contains a review and self evaluation section challenge problems and answers and solutions for all exercises in the study guide Foundations of College Chemistry is the book that defined the prep intro chemistry market 35 years ago And it's been a market leader ever since Hein Arena is known for its accuracy clear no nonsense approach and direct writing style Strong problem solving and carefully constructed problem sets make this book a standout among its many imitators

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Table of Contents Periodic Table Basic Speedy Study Guides

1. Understanding the eBook Periodic Table Basic Speedy Study Guides
 - The Rise of Digital Reading Periodic Table Basic Speedy Study Guides
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Basic Speedy Study Guides
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Basic Speedy Study Guides
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Basic Speedy Study Guides
 - Personalized Recommendations
 - Periodic Table Basic Speedy Study Guides User Reviews and Ratings
 - Periodic Table Basic Speedy Study Guides and Bestseller Lists
5. Accessing Periodic Table Basic Speedy Study Guides Free and Paid eBooks
 - Periodic Table Basic Speedy Study Guides Public Domain eBooks
 - Periodic Table Basic Speedy Study Guides eBook Subscription Services
 - Periodic Table Basic Speedy Study Guides Budget-Friendly Options
6. Navigating Periodic Table Basic Speedy Study Guides eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Basic Speedy Study Guides Compatibility with Devices
 - Periodic Table Basic Speedy Study Guides Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Basic Speedy Study Guides
 - Highlighting and Note-Taking Periodic Table Basic Speedy Study Guides
 - Interactive Elements Periodic Table Basic Speedy Study Guides
8. Staying Engaged with Periodic Table Basic Speedy Study Guides

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Periodic Table Basic Speedy Study Guides
- 9. Balancing eBooks and Physical Books Periodic Table Basic Speedy Study Guides
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Basic Speedy Study Guides
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Basic Speedy Study Guides
 - Setting Reading Goals Periodic Table Basic Speedy Study Guides
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Basic Speedy Study Guides
 - Fact-Checking eBook Content of Periodic Table Basic Speedy Study Guides
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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