

PERIODIC TABLE

(BASIC)

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H Hydrogen 1.008	2 He Helium 4.003																
3 Li Lithium 6.941	4 Be Beryllium 9.012															19 K Potassium 39.098	20 Ca Calcium 40.078
5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180											21 Sc Scandium 44.956	22 Ti Titanium 47.88
11 Na Sodium 22.990	12 Mg Magnesium 24.305	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.45	18 Ar Argon 39.948	29 Cu Copper 63.546	30 Zn Zinc 65.38	31 Ga Gallium 69.723	32 Ge Germanium 72.63	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80	37 Rb Rubidium 85.468	38 Sr Strontium 87.62
39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.906	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.91	46 Pd Palladium 106.36	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.757	52 Te Tellurium 127.6	53 I Iodine 126.905	54 Xe Xenon 131.29	55 Ba Barium 137.327	56 La Lanthanum 138.905
57 Pr Praseodymium 140.908	58 Ce Cerium 140.12	59 Th Thorium 232.0377	60 Pa Protactinium 231.03688	61 Ac Actinium 227.0277	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.259	69 Tm Thulium 168.934	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.967	72 Hf Hafnium 178.49	73 Ta Tantalum 180.94788	74 W Tungsten 183.84
75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium 209	85 At Astatine 210	86 Rn Radon 222	87 Fr Francium 223	88 Ra Radium 226	89 Ac Actinium 227	90 Th Thorium 232.0377	91 Pa Protactinium 231.03688	92 U Uranium 238.02891
93 Np Neptunium 237.04817	94 Pu Plutonium 244.0642	95 Am Americium 243.06136	96 Cm Curium 247.07645	97 Bk Berkelium 247.07125	98 Cf Californium 251.0833	99 Es Einsteinium 252.0832	100 Fm Fermium 257.10528	101 Md Mendelevium 258.10528	102 No Nobelium 259.10528	103 Lr Lawrencium 262.10528	104 Rf Rutherfordium 261.10528	105 Db Dubnium 262.10528	106 Sg Seaborgium 266.10528	107 Bh Bohrium 264.10528	108 Hs Hassium 277.10528	109 Mt Meitnerium 268.10528	110 Ds Darmstadtium 271.10528
111 Rg Roentgenium 281.10528	112 Cn Copernicium 285.10528	113 Nh Nihonium 286.10528	114 Fl Flerovium 289.10528	115 Mc Moscovium 290.10528	116 Lv Livermorium 293.10528	117 Ts Tennessine 294.10528	118 Og Oganesson 294.10528	119 Uut Ununtrium 295.10528	120 Uuq Ununquadium 299.10528	121 Uub Unbium 301.10528	122 Uus Ununseptium 303.10528	123 Uuh Ununhexium 305.10528	124 Uuo Ununoctium 307.10528	125 Uuq Ununquadium 309.10528	126 Uus Ununseptium 311.10528	127 Uuh Ununhexium 313.10528	128 Uuo Ununoctium 315.10528

For elements with no stable isotopes, the mass number of the isotope with the longest half-life is in parentheses.

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Hydrogen

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic forms (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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