

Periodic Table of the Elements

This table gives information about the chemical elements. Elements are grouped into eight classes according to their properties. Each class is shown in a different color. Hydrogen does not belong to any one class.

Atomic number: 6

Chemical symbol: C

Element name: Carbon

Atomic weight: 12.011

Number of electrons in each shell: 2, 4

(3 indicates atomic weight of the most stable isotope)

Alkali metals Alkaline earth metals Transition metals Lanthanide series Actinide series Other metals Nonmetals Noble gases																		12.011 1 indicates atomic weight of the most stable isotope									
</																											

Note: Elements 112, 113, and 117 are not currently known.

Note: The adjacent numbers to those reported in tables by International Union of Pure and Applied Chemistry (IUPAC) are the accepted values in the field of chemistry. It is the business of scientists and engineers to know.

57 La Lanthanum 138.9055	58 Ce Cerium 140.12	59 Pr Praseodymium 140.90768	60 Nd Neodymium 144.24	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92534	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93033	68 Er Erbium 167.26	69 Tm Thulium 168.93421	70 Yb Ytterbium 173.054	71 Lu Lutetium 174.967
89 Ac Actinium (227)	90 Th Thorium 232.0377	91 Pa Protactinium 231.03688	92 U Uranium 238.02891	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)

Periodic Table Of The Elements Cheap Chart Cheap Charts

Tao Wei



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