

equivalent Fractions Math Task Cards

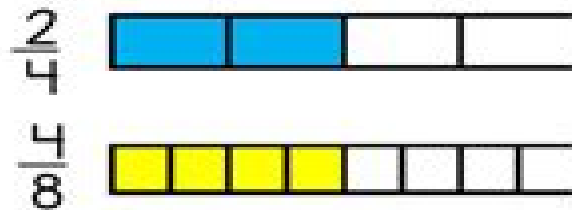
equivalent fractions

Find the missing number to make these fractions equivalent.

$$\frac{2}{3} = \frac{\boxed{}}{12}$$

equivalent fractions

Use the model to help you answer the question. Are these fractions equivalent?



equivalent fractions

Fill in the chart below:

Model	Fraction	Whole Number
		
	$\frac{3}{3}$	

3

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equivalent fractions

Find the missing number to make these fractions equivalent

$$\frac{4}{7} = \frac{\boxed{}}{28}$$

4

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Performance Task 3rd Grade Math Fractions

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web oxidation reduction redox reactions oxidation $\text{Cu} + \text{Cu}^{2+} + 2\text{e}^-$ reduction $\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$ l when the half reactions listed above are combined into a balanced chemical equation the ratio of the coefficients $\text{Cu} : \text{MnO}_4^- : 4\text{H}^+ : 3\text{e}^-$ is

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web solution 1 the two half reactions balanced as if in acidic solution $2\text{CN}^- + \text{Au} \rightarrow 2\text{e}^- + 2\text{H}^+ + \text{O}_2 + 2\text{H}_2\text{O}$ 2 make electrons equal convert to basic solution $4\text{CN}^- + 2\text{Au} \rightarrow 2\text{e}^- + 2\text{H}_2\text{O} + \text{O}_2 + 2\text{OH}^-$ 3 the final answer

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web the half equation method separates the oxidation and reduction of a redox reaction in half reactions overall scheme for the half reaction method step 1 split reaction into half reactions reduction and oxidation step 2 balance the charge or oxidation number with electrons step 3 balance c e o by adding c e h₂o

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bromate v ions

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