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side there are 2 also unbalanced chlorine but they both come from the same compound so this becomes really simple to
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 $4\text{H}_2 + 2\text{O}_2 \rightarrow 4\text{H}_2\text{O}$
 $5\text{Pb} + 2\text{HCl} \rightarrow 2\text{H}_2 + 5\text{PbCl}_2$
 $6\text{Al} + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KAl} + 3\text{S} + 4\text{O}_2$
 $1\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$

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