

2025 Tax Rate	Single	Married Filing Jointly	Married Filing Separately	Head of Household
10%	\$0 to \$11,925	\$0 to \$23,850	\$0 to \$11,925	\$0 to \$17,000
12%	\$11,926 to \$48,475	\$23,851 to \$96,950	\$11,926 to \$48,475	\$17,001 to \$64,850
22%	\$48,476 to \$103,350	\$96,951 to \$206,700	\$48,476 to \$103,350	\$64,851 to \$103,350
24%	\$103,351 to \$197,300	\$206,701 to \$394,600	\$103,351 to \$197,300	\$103,351 to \$197,300
32%	\$197,301 to \$250,525	\$394,601 to \$501,050	\$197,301 to \$250,525	\$197,301 to \$250,500
35%	\$250,526 to \$626,350	\$501,051 to \$751,600	\$250,526 to \$375,800	\$250,501 to \$626,350
37%	\$626,351 or greater	\$751,601 or greater	\$375,801 or greater	\$626,351 or greater

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web balancing chemical equations worksheet answer key balance the equations below

$$\begin{array}{l} 1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3 \\ 3 \text{ Fe} + 2 \text{ O}_2 \rightarrow \text{Fe}_3\text{O}_4 \\ 2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2 \\ 2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2 \\ 4 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O} \\ 5 \text{ Pb} + 2 \text{ OH}^- + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 \\ 6 \text{ Al} + 3 \text{ Br}_2 \rightarrow 3 \text{ AlBr}_3 \\ 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 \rightarrow 3 \text{ H}_2\text{O} + 2 \text{ CO}_2 + 4 \text{ H}_2\text{O} + 8 \text{ HCl} + 3 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 + 3 \text{ CO}_2 + 4 \text{ CH}_4 \end{array}$$

5 1 1 practice problems writing and balancing chemical equations - May 05 2023

web may 28 2020 write a balanced chemical equation for each step of the process the first step is the decomposition of solid calcium carbonate from seashells to form solid calcium oxide and gaseous carbon dioxide the second step is the formation of solid calcium hydroxide as the only product from the reaction of the solid calcium oxide with liquid water