

Key to constructing the periodic table of Martian Elements

Elements List:

A	I	Q	Y
B	J	R	Z
C	K	S	@
D	L	T	#
E	M	U	%
F	N	V	&
G	O	W	
H	P	X	



Guide:

1. The most metallic element is X.
2. The element I is the lightest element with 7 valence electrons.
3. The gases with full outer shells are B, L, W, and J. B is the lightest, W is the heaviest, and L is in period 2.
4. The lightest element of all is A.
5. All the following elements have 3 energy shells and the number of electrons listed in the outermost shell is as follows: &-1, Z-2, Y-3, H-4, M-5, O-6, K-7.
6. Element H has 14 protons.
7. C forms cations with a charge of +1 and has an atomic weight of 5.
8. Q has only one electron in its outer shell but has 4 shells.
9. The elements E, Y, R and # all have the same valence electron arrangement and like to form ions by losing three electrons. E is the lightest followed by Y, R, and # in increasing weight.
10. % is the heaviest of all elements and is radioactive.
11. P will form ionic compounds with the formula PI_2 and is the largest member of its group.
12. D is in period 2 and forms salts with for formula DO.
13. F has the same number of protons as our carbon and is in the same family as H, T, and %.
14. G will form covalent compounds with I and O but is not in the same family as either of them. G also has the highest ionization energy of its group.
15. The martian solvent for our most important liquid A_2N .
16. Martian ionic compounds include: AV, @I₂, DU, @U, and @₃S₂.
17. The following elements are diatomic: I, K and V.

Every space should be filled. Write the correct atomic number in each box. You can make up Martian names for all the elements if you want. . . be creative!

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