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 $\text{aq C}_3\text{H}_7\text{NO}_2 + \text{aq H}_3\text{O}^+ \rightleftharpoons \text{aq C}_3\text{H}_8\text{NO}_2 + \text{aq H}_2\text{O}$ l $\text{CH}_3\text{CO}_2\text{H} + \text{aq NH}_3 \rightleftharpoons \text{aq CH}_3\text{CO}_2^- + \text{aq NH}_4^+$ aq $\text{SbF}_5 + \text{aq 2HF} \rightleftharpoons \text{aq SbF}_6^- + \text{aq 2H}^+$ identify the conjugate acid base

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