

Lab 01: Measurement and the Standard Curve

Experiment 1 Standard Curve

Table 1

Tube	Final Protein Concentration	μL of BSA protein stock (0.5 mg/mL) needed	Bradford Reagent	0.15M NaCl for 1.0 mL final volume
8	50 $\mu\text{g/mL}$	100 μL	900 μL	0 μL
7	25 $\mu\text{g/mL}$	50 μL	900 μL	50 μL
6	15 $\mu\text{g/mL}$	30 μL	900 μL	70 μL
5	10 $\mu\text{g/mL}$	20 μL	900 μL	80 μL
4	7.5 $\mu\text{g/mL}$	15 μL	900 μL	85 μL
3	5.0 $\mu\text{g/mL}$	10 μL	900 μL	90 μL
2	2.5 $\mu\text{g/mL}$	5 μL	900 μL	95 μL
1	0 $\mu\text{g/mL}$	0 μL	900 μL	100 μL

Table 2

Final Concentration	Absorbance 595nm
50 $\mu\text{g/mL}$	0.91
25 $\mu\text{g/mL}$	0.58
15 $\mu\text{g/mL}$	0.43
10 $\mu\text{g/mL}$	0.30
7.5 $\mu\text{g/mL}$	0.23
5.0 $\mu\text{g/mL}$	0.15
2.5 $\mu\text{g/mL}$	0.04
0 $\mu\text{g/mL}$	0.06

Part B: Protein Concentration in Milk

Table 3

My Milk Type = 2% _____ The Dilution of milk I used was: 1:10 _____

Tube	Unknown Volume	Bradford Reagent	0.15M NaCl for 1.0 mL final	A595nm	Protein Concentration Measured in cuvette ($\mu\text{g/mL}$)	Protein Concentration in Unknown Volume ($\mu\text{g/mL}$)	Accounting for dilution (multiply by dilution factor)	(Original) Concentration of Unknown ($\mu\text{g}/\mu\text{L}$)
1	5 μL	900 μL	95	0.37 nm	13.71 $\mu\text{g/mL}$	2,742 $\mu\text{g/mL}$	27,420	27.42 $\mu\text{g/mL}$
2	10 μL	900 μL	90	0.26 nm	25.42 $\mu\text{g/mL}$	2,542 $\mu\text{g/mL}$	25,420	25.42 $\mu\text{g/mL}$
3	20 μL	900 μL	80	0.99 nm	33.11 $\mu\text{g/mL}$	1,655.40 $\mu\text{g/mL}$	16,554	16.55 $\mu\text{g/mL}$
4	40 μL	900 μL	60	1.26 nm	42.14 $\mu\text{g/mL}$	1,053.50 $\mu\text{g/mL}$	10,535	10.54 $\mu\text{g/mL}$

- a. What was your final R^2 value? What does this tell us about the precision of your measurements?

0.98

This tells us it is around 96% linear

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Thomas Griffiths



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