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Thomas Tanner



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which decides the nature of the above solution is 1 0 25m 2 0 5m 3 0 888m 4 none of

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$$m = \frac{1000 \times M}{1000d - M}$$

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web 1 molarity $m = \frac{m}{M}$ mass molar mass $x = \frac{0.4500 \text{ L} \times 0.825 \text{ g/L} \times 0.0129 \text{ mol/L}}{0.825 \text{ g/L} \times 0.00581158 \text{ mol/L} + 0.4500 \text{ kg} \times 0.0129 \text{ mol/L}}$ 3 mole fraction $n_2 = \frac{0.825 \text{ g/L} \times 0.00581158 \text{ mol/L}}{0.825 \text{ g/L} \times 0.00581158 \text{ mol/L} + 0.4500 \text{ g} \times 0.0129 \text{ mol/L}}$ 24 97918401 mol

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$$m_{\text{mix}} = \frac{m_1 v_1 + m_2 v_2}{v_1 + v_2}$$

$$1.5 \times 480 + 1.2 \times 520 = 480 + 520$$

$$1.344 m$$
 q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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