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2 cu s zn 2 10 2 1 10 2 1 z n s c u 2

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when the oxidized species is separated from the

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l o r t n f l n q choose 1 answer choose 1 answer choice a at equilibrium e c e l l calculates to a value of 0 and the battery is

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cu 2 ions have greater tendency to

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to zn 2 ions in a cell with a zinc electrode and a

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kj mol r is expressed as j mol k so we can convert r or g o to match units let s plug in 300 000 j for g o to match r divide 300
000 by

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voltage produced by the cell in ii above when nitrate

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calculate the standard gibbs

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determine what is oxidized and what is reduced step 3

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