

# Chemistry Reference Tables

| Name                               | Value                                                                                                                                   |
|------------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------|
| Avogadro's number                  | $6.022 \times 10^{23}$ particles/mole                                                                                                   |
| Gas constant ( <i>R</i> )          | $0.0821 \frac{\text{L atm}}{\text{mole K}}$<br>$62.4 \frac{\text{L mmHg}}{\text{mole K}}$<br>$8.314 \frac{\text{L kPa}}{\text{mole K}}$ |
| Standard pressure                  | 1.00 atm = 101.3 kPa =<br>760. mmHg = 760. torr                                                                                         |
| Standard temperature               | 0°C or 273K                                                                                                                             |
| Volume of 1 mole of any gas at STP | 22.4 L                                                                                                                                  |

| Thermodynamic Constants       | Symbol        | Value                                                                                                                                                                             |
|-------------------------------|---------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| Heat of fusion of water       | $H_f$ (water) | 334 J/g                                                                                                                                                                           |
| Heat of vaporization of water | $H_v$ (water) | 2,260 J/g                                                                                                                                                                         |
| Specific heat of water        | $C_p$ (water) | $2.05 \frac{\text{J}}{\text{g}^\circ\text{C}}$ for ice, $2.02 \frac{\text{J}}{\text{g}^\circ\text{C}}$<br>for steam,<br>$4.18 \frac{\text{J}}{\text{g}^\circ\text{C}}$ for liquid |

| Metal     | Specific Heat $\frac{\text{J}}{\text{g}^\circ\text{C}}$ | Density (g/cm <sup>3</sup> ) | Melting Point (°C) |
|-----------|---------------------------------------------------------|------------------------------|--------------------|
| Aluminum  | 0.897                                                   | 2.702                        | 660                |
| Copper    | 0.385                                                   | 8.92                         | 1083               |
| Gold      | 0.129                                                   | 19.31                        | 1064               |
| Iron      | 0.449                                                   | 7.86                         | 1535               |
| Lead      | 0.129                                                   | 11.3437                      | 328                |
| Magnesium | 1.023                                                   | 1.74                         | 649                |
| Mercury   | 0.140                                                   | 13.5939                      | -39                |
| Nickel    | 0.444                                                   | 8.90                         | 1455               |
| Titanium  | 0.523                                                   | 4.5                          | 1660               |
| Zinc      | 0.388                                                   | 7.14                         | 420                |

# Ncdpi Quick Reference Guides

**Ensheng Dong**



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