

Class 12 Chemistry Notes PDF

Solution

Solution :- A solution is a homogeneous mixture of two or more substances, in which the solute is uniformly dispersed at the molecular or ionic level within the solvent.

In solution solute(s) can be more than one but solvent must be unique.

Homogeneous mixture :-

That mixture which is found as uniform is called homogeneous mixture.

Heterogeneous mixture :-

That mixture where though out the solution the composition is not uniform.

Solute :-

The substance that is dissolved in a solvent to form a solution. It is present in smaller amount.

Kinetics

Chemistry which deals with the rate of chemical reactions, along the rate of reaction and by which the reaction proceed called kinetics.

The rate of reaction is the concentration of any one of reactant or product per unit time.

$$= -\frac{\Delta[R]}{\Delta t} = +\frac{\Delta[P]}{\Delta t}$$

negative and positive sign :-

- The rate of reaction is always positive.
- The minus sign along with first term is used to show that the concentration of reactant is decreasing.

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Lingsheng Yao



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