

SUMMATIVE TEST

Science quarter 4

NAME:

SECTION:

- 1. Which of the following characteristics best describes gas?*
 - a. Gases are lighter than air.*
 - b. Gases have high densities.*
 - c. Gases exhibit similar chemical behavior.*
 - d. Gases do not have definite shape and volume.*

- 2. How are gas particles described according to the Kinetic Molecular Theory?*
 - a. Gases have mass.*
 - b. Gas particles move rapidly in straight lines, travel constantly, and in a random direction.*
 - c. Gas particles collide with each other and with the walls of the container in a perfectly elastic manner.*
 - d. All of the above*

- 3. Which of the following changes will result to a decrease in gas pressure?*
 - a. Adding more gas particles*
 - b. Increasing the temperature*
 - c. Liquefying the gas particles*
 - d. Increasing the volume of the container*

- 4. Which gas law describes the relationship between pressure and volume at constant temperature?*
 - a. Avogadro's Law*
 - b. Boyle's Law*
 - c. Charles' Law*
 - d. Gay-Lussac's Law*

- 5. Which of the following is not equal to 1.00 atm of pressure?*
 - a. 760 cm Hg*
 - b. 760 mm Hg*
 - c. 760 torr*
 - d. 101,325 Pa*

- 6. Which gas law is a combination of all gas laws?*
 - a. Boyle's Law*
 - b. Charles' Law*
 - c. Gay-Lussac's Law*
 - d. Ideal gas Law*

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