



Muller Martini Bravo Manual

Tao Wei



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 molecules and ions contain polar bonds which of these molecules and ions have dipole moments a ClF_5 b ClO_2 c TeCl_2 d
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web the relative electronegativities of the bonded atoms is $\text{H} < \text{C} < \text{Cl}$ and so the bond moments all point toward the Cl end of the
 molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry
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